

CORRECTIONS

CHARACTERIZATION OF BOVINE URINE AND ADIPOSE INTERLABORATORY PERFORMANCE EVALUATION SAMPLES CONTAINING BIOLOGICALLY INCORPORATED CHLOROPHENOLS, by Paul J. Marsden, E. Neil Amick, Fred L. Shore, Llewellyn R. Williams, Verle R. Bohman,* and Clifton R. Blincoe, *J. Agric. Food Chem.* 1986, 34, 795-800.

Page 795: Author Blincoe should have read Clifton R. Blincoe. Reference to C.R.B. should be University of Nevada—Reno, Reno, NV.

PRODUCTION OF OXALIC ACID FROM DRY POWDER OF *PARTHENIUM HYSTEROPHORUS* L., by Jyoti D. Mane, Sadashiv J. Jadhav,* and Nanduri A. Ramaiah, *J. Agric. Food Chem.* 1986, 34, 989-990.

A portion of the Materials and Methods section was inadvertently omitted.

MATERIALS AND METHODS

Whole dry plants were collected and ground into a powder, which was used as raw material. The reaction was carried out in a three-neck flask containing a mixture of plant material (10 g) and catalyst (5 mg). Initially, the flask was ice cooled, and an appropriate quantity of pre-cooled concentrated nitric acid was added into it with slow stirring. After 30 min, a predetermined quantity of cold concentrated sulfuric acid was added. Subsequently, the temperature of the reaction mixture was maintained at a desired temperature for specified time. The resultant mixture was then filtered through a nylon mesh. The filtrate was assayed for oxalic acid content by a procedure that involved the precipitation of oxalic acid with calcium chloride, followed by recovery of the calcium oxalate by filtration, subsequent dissolution in dilute sulfuric acid, and then titration with standard permanganate (0.1 N). The quantity of oxalic acid was calculated by the following formula:

$$\text{oxalic acid (g/10 g dry powder)} = \frac{0.1 \times V \times R \times DF \times 45}{S \times 1000}$$

where V = volume of the filtrate (mL), R = titration reading (mL), DF = dilution factor (16), and S = volume of filtrate diluted (16 times) for oxalate precipitation (10 mL). Oxalic acid was recovered from the filtrate by crystallization induced by cooling.